

Rutherford Backscattering Spectrometry

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Geiger-Marsden Experiment



Top: Expected results: alpha particles passing through the plum pudding model of the atom undisturbed.

Bottom: Observed results: a small portion of the particles were deflected, indicating a small, concentrated positive charge.

Rutherford Backscattering Spectrometry



RBS is an analytical technique where high energy ions (~2 MeV) are scattered from atomic nuclei in a sample. The energy of the back-scattered ions can be measured to give information on sample composition as a function of depth.

Van de Graaff accelerator



http://archive.thedailystar.net/newDesign/print_news.php?nid=73473



Rutherford Backscattering Spectrometry

3 MeV Pelletron accelerator



beam size ϕ 1-3 mm flat sample can be rotated



Ι

NEC Pelletron



- Ionization chamber
- Acceleration tube
- Focusing quadrupole
- Steering magnet
- RBS end station



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Primary Beam Energy



thin film projected on to a plane: atoms/cm²

 $(Nt)[at/cm^2] = N[at/cm^3] * t[cm]$

Figure after W.-K. Chu, J. W. Mayer, and M.-A. Nicolet, *Backscattering Spectrometry* (Academic Press, New York, 1978).

Elastic Two-Body Collision



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Rutherford Scattering Cross Section





Coulomb interaction between the nuclei: exact expression -> quantitative method

$$\sigma_{R}(E,\theta) \propto \left(\frac{Z_{1}Z_{2}}{4E}\right)^{2} \left[\sin^{-4}\left(\frac{\theta}{2}\right) - 2\left(\frac{M_{1}}{M_{2}}\right)\right] \propto \left(\frac{Z_{2}}{E}\right)^{2}$$

Electron Stopping



Figure after W.-K. Chu, J. W. Mayer, and M.-A. Nicolet, *Backscattering Spectrometry* (Academic Press, New York, 1978).

RBS – Simulated Spectra

hypothetical alloy $Au_{0.2}In_{0.2}Ti_{0.2}AI_{0.2}O_{0.2}/C$

Element (Z,M): O(8,16), Al(13,27), Ti(22,48), In(49,115), Au(79,197)





Calibration Sample



SIMNRA Simulation Program for RBS and ERD



Ι

Cu-Nb-W Alloy on SiO₂/Si



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Courtesy N. Vo and R.S. Averback

Thickness Effects



Incident Angle Effects



Surface peaks do not change position with incident angle



I. Petrov, P. Losbichler, J. E. Greene, W.-D. Münz, T. Hurkmans, and T. Trinh, *Thin Solid Films*, 302 179 (1997)

RBS: Oxidation Behavior

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Experimental spectra and simulated spectra by RUMP

- Free-standing polyamide films are too thin to give sufficient signal in the RBS.
- Use the added stopping power of the polymer to split the Pt peak in the RBS spectrum.



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Areal mass density by RBS



JM Dennison, X Xie, CJ Murphy, DG Cahill - ACS Applied Nano Materials, 2018

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Additional Analytical Capabilities

Elastic Recoil Detection



Additional Analytical Capabilities

Nuclear Reaction Analysis



Courtesy of E.J. Cho, N. Perry, University of Illinois

Ion Irradiation





SNICS source

He source

Irradiation end station

RBS Summary



- Quantitative technique for elemental composition
- Requires flat samples; beam size ϕ 1-3 mm
- Non-destructive
- Detection limit varies from 0.1 to 10⁻⁶, depending on Z
 - •optimum for heavy elements in/on light matrix, e.g. Ta/Si, Au/C...
- \bullet Depth information from monolayers to 1 μm



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