

An electron is confined to a 1-dimensional infinite potential well of width L . The energy of the electron in the first excited state ($n = 2$) is 12 eV.

1. [5 points] What is the energy of the electron in its ground state (e.g., its lowest energy state)?

$$E_{n=1} = E_{n=2} / 4 = 3 \text{ eV}$$

2. [5 points] What is the width L of the well?

$$L = 1/(\sqrt{8 m E_1})/\hbar = .354 \text{ nm}$$

3. [6 points] What is the minimum energy photon that could be absorbed so that the electron can transition from the first to the fourth ($n = 5$) excited state?

$$E = (E_5 - E_2) = (5^2 - 2^2) 3 \text{ eV} = 63 \text{ eV}$$

4. [4 points] If we double the width of the well what happens to the energy of the first excited state?

It goes down by a factor of 4.