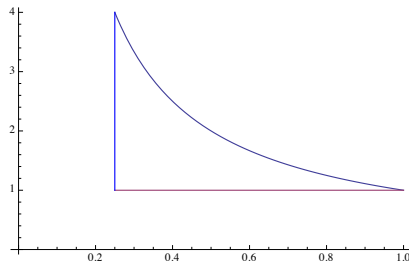


1. [5 points] How much heat is required to heat six moles of O_2 gas from 300 K to 375 K at a constant volume of 0.2 m^3 ?

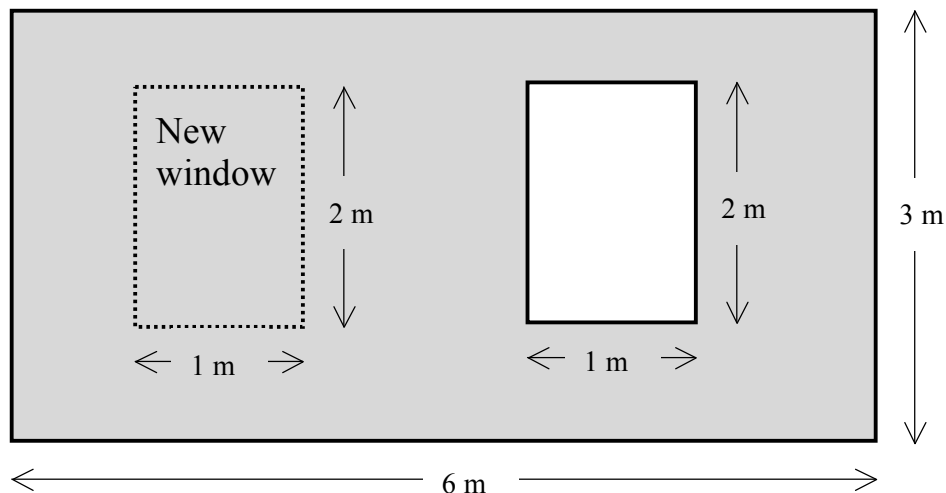
$$C_V = \frac{5}{2} nR \rightarrow Q = \frac{5}{2} n R (75 \text{ K}) = 9353 \text{ J}$$

2. [5 points] Draw (roughly) the P-V diagram for the following 3 process of an ideal gas which form a closed cycle: isothermal compression, constant volume cooling, constant pressure heating.



With axes labeled and a counter clockwise arrow!

2. [10 points] A homeowner wants to add a new window to one of the walls in their house. The wall already has a window and the second window is planned to be identical in size:



The wall is 3 m high and 6 m wide, while the window is 2 m high and 1 m wide, as shown. The walls are 15cm thick and the glass is 1cm thick. On a cold day in Chicago, the outside temperature is -14°C , and the homeowner wants to keep the temperature inside his house at $+20^\circ \text{C}$. Suppose the thermal conductivity of the solid part of the wall is 0.30 W/mK and the thermal conductivity of the window is 4.50 W/mK , what is the ratio, $H_{\text{new}}/H_{\text{old}}$, of the new heat loss to the old heat loss through this wall? Explain your reasoning.

$$H = dT \left(\frac{k_{\text{wall}} A_{\text{wall}}}{d_{\text{wall}}} + \frac{k_{\text{window}} A_{\text{window}}}{d_{\text{window}}} \right) \rightarrow H_{\text{new}}/H_{\text{old}} = 1.961$$