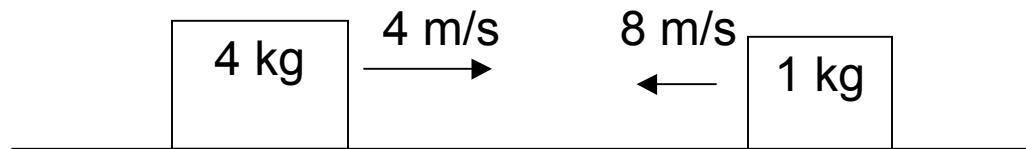


**Physics 213****Quiz 1-1 [20 points]** Name \_\_\_\_\_

1. [11 points]



A 4-kg block moving at 4 m/s collides with a 1-kg block moving at 8 m/s in the opposite direction on a frictionless floor. After the collision, the blocks stick together and move as a single unit.

a) [3 points] What are the kinetic energies of the blocks before collision?

$$\frac{1}{2} m v_1^2 + \frac{1}{2} m v_2^2 = 32 \text{ J} + 32 \text{ J}$$

b) [5 points] What is the kinetic energy of the combined block after the collision?

$$p_{\text{before}} = p_{\text{after}} = (4 \cdot 4 - 8) = 8 \text{ kg m/s to the right}$$

$$E_{\text{after}} = p^2 / (2m) = 8^2 / (2 \cdot 5) = 6.4 \text{ J}$$

c) [3 points] What is the thermal energy released in the collision?

$$E_{\text{thermal}} = E_{\text{KE, before}} - E_{\text{KE, after}} = 57.6 \text{ J}$$

2. [9 points]

a) [3 points] The pressure of an ideal diatomic gas is isothermally increased by 25%. By what factor does the average rotational energy of a molecule increase? Explain your reasoning.

*\*1. Since the temperature stays the same equipartition tells us that the average energy of a single molecule stays the same.*

b) [3 points] Which of the following equations is always correct for an *isothermal* process of an ideal gas? (Circle the correct statements.)

X i)  $(PV)_{\text{before}} = (PV)_{\text{after}}$

ii)  $Q = C_v \Delta T$

iii)  $Q = 0$

iv)  $P_{\text{before}} = P_{\text{after}}$

X v)  $U_{\text{before}} = U_{\text{after}}$

vi)  $Q = W_{\text{on}}$

c) [3 points] Which of the following equations is always correct for an *adiabatic* process of an ideal gas? (Circle the correct statements.)

i)  $(PV)_{\text{before}} = (PV)_{\text{after}}$

ii)  $Q = C_v \Delta T$

X iii)  $Q = 0$

iv)  $P_{\text{before}} = P_{\text{after}}$

v)  $U_{\text{before}} = U_{\text{after}}$

vi)  $Q = W_{\text{on}}$