SMSpring96B

A large number of monatomic gas atoms of mass m are confined to a cubic box of volume $V=L^3$. One wall of the box of area $A=L^2$ adsorbs atoms with a

binding energy ϕ . The atoms are otherwise free to move and may be treated as non-interacting indistinguishable quantum particles, both on the surface and in the gas. The temperature T is constant throughout the system.

of temperature where the effects of Fermi and Bose-Einstein statistics can be ignored.

(a) Write down the condition which ensures that the gas atoms are in a regime

In the following, assume that the gas is always in the regime specified in part (a). (b) Calculate the partition function for N_g particles in the gas, expressing

your answer in terms of $N_{\rm g}$, V, and $\gamma \equiv (2m~{\rm k_BT/\hbar^2})^{1/2}$. (c) Calculate the partition function for $N_{\rm s}$ particles on the surface, also

assuming it is a large number and giving the answer in terms of

 N_s , A, ϕ , and γ . (d) From parts (b) and (c), find expressions for the chemical potentials for the atoms in the gas μ_g and on the surface μ_s .

(e) Calculate n_s , the density per unit area of atoms on the surface in equilibrium at temperature T, when there are n_g atoms per unit volume in the gas. Give an expression in terms of the quantities defined above.