

An idealized model of a polymer consists of a chain of  $N$  links lying in the  $xy$ -plane. Each of the links has length  $a$  and is free to point along each of the four lattice directions in the plane, as shown in the figure. One end of the chain is fixed to the origin  $O$ , and the other end is located at the point  $\mathbf{L} = (L_x, L_y)$ . Links are allowed to double back or form closed loops, as shown in the figure. The total energy  $U$  of the configuration is the sum of link energies: zero for each link lying parallel to the  $x$  axis and  $\epsilon$  ( $> 0$ ) for each link lying parallel to the  $y$  axis.  $N_x^+$ ,  $N_x^-$ ,  $N_y^+$  and  $N_y^-$  are, respectively, the number of links whose heads point in the positive or negative  $x$  or  $y$  directions, relative to the previous link.

- (a) The exact expression for the number of possible configurations  $\Omega$  for given  $N$ ,  $N_x^+$ ,  $N_x^-$ ,  $N_y^+$  and  $N_y^-$ , is given by

$$\Omega = \frac{N!}{N_x^+! N_x^-! N_y^+! N_y^-!}$$

Use Stirling's approximation to the factorial function [i.e.,  $\ln M! \approx M(\ln M - 1)$ ] to determine the entropy  $S$  as a function of  $N$ ,  $N_x^+$ ,  $N_x^-$ ,  $N_y^+$  and  $N_y^-$ .

- (b) Express  $N_x^+$ ,  $N_x^-$ ,  $N_y^+$  and  $N_y^-$ , and hence  $S$ , in terms of  $N$ ,  $L_x$ ,  $L_y$  and  $U$ .

For the present system the first law of thermodynamics reads

$$dU = T dS + \mathbf{I} \cdot d\mathbf{L},$$

where  $T$  is the temperature and  $\mathbf{I} = (I_x, I_y)$  is the tension applied to the end of the polymer.

- (c) Express the components of  $\mathbf{I}$  in terms of derivatives of  $S$ . Hence, determine  $\mathbf{I}$  explicitly in terms of  $N$ ,  $L_x$ ,  $L_y$  and  $U$ . Determine  $L_y$  for the situation that  $I_y = 0$ .
- (d) Determine  $T$  in terms of  $N$ ,  $L_x$ ,  $L_y$  and  $U$ .
- (e) Explain briefly the origin of the formula for  $\Omega$  given in part (a).

