



Rutherford Backscattering Spectrometry

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WITHOUT



Geiger-Marsden Experiment



Top: Expected results: alpha particles passing through the plum pudding model of the atom undisturbed.

Bottom: Observed results: a small portion of the particles were deflected, indicating a small, concentrated positive charge.

Rutherford Backscattering Spectrometry



RBS is an analytical technique where high energy ions (~2 MeV) are scattered from atomic nuclei in a sample. The energy of the back-scattered ions can be measured to give information on sample composition as a function of depth.

Van de Graaff accelerator



http://archive.thedailystar.net/newDesign/print_news.php?nid=73473



Rutherford Backscattering Spectrometry

3 MeV Van de Graaff accelerator



beam size ϕ 1-3 mm flat sample can be rotated

Rutherford Backscattering Spectrometry

3 MeV Pelletron accelerator



beam size ϕ 1-3 mm flat sample can be rotated



NEC Pelletron

- Ionization chamber
- Acceleration tube
- Focusing quadrupole
- Steering magnet
- RBS end station



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Primary Beam Energy



thin film projected on to a plane: atoms/cm²

 $(Nt)[at/cm^2] = N[at/cm^3] * t[cm]$

Figure after W.-K. Chu, J. W. Mayer, and M.-A. Nicolet, *Backscattering Spectrometry* (Academic Press, New York, 1978).



Elastic Two-Body Collision



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Rutherford Scattering Cross Section





Coulomb interaction between the nuclei: exact expression -> quantitative method

$$\sigma_R(E,\theta) \propto \left(\frac{Z_1 Z_2}{4E}\right)^2 \left[\sin^{-4}\left(\frac{\theta}{2}\right) - 2\left(\frac{M_1}{M_2}\right)\right] \propto \left(\frac{Z_2}{E}\right)^2$$



Electron Stopping



Figure after W.-K. Chu, J. W. Mayer, and M.-A. Nicolet, *Backscattering Spectrometry* (Academic Press, New York, 1978).

RBS – Simulated Spectra

hypothetical alloy $Au_{0.2}In_{0.2}Ti_{0.2}AI_{0.2}O_{0.2}/C$

Element (Z,M): O(8,16), Al(13,27), Ti(22,48), In(49,115), Au(79,197)



SIMNRA Simulation Program for RBS and ERD





Calibration Sample





Cu-Nb-W Alloy on SiO₂/Si



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Courtesy N. Vo and R.S. Averback

Thickness Effects







<mark>כ</mark>

Incident Angle Effects



Surface peaks do not change position with incident angle

Example: Average Composition







RBS: Oxidation Behavior



TiN/SiO₂

As-deposited



Experimental spectra and simulated spectra by RUMP



Areal mass density by RBS

 Free-standing polyamide films are too thin to give sufficient signal in the RBS.

 Use the added stopping power of the polymer to split the Pt peak in the RBS spectrum.



Areal mass density by RBS





- Quantitative technique for elemental composition
- Requires flat samples; beam size ϕ 1-3 mm
- Non-destructive
- Detection limit varies from 0.1 to 10⁻⁶, depending on Z
 - •optimum for heavy elements in/on light matrix, e.g. Ta/Si, Au/C...
- \bullet Depth information from monolayers to 1 μm



Optimizing Simultaneous PIXE and RBS Capabilities

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