

**Syllabus**  
**MSE 307: Materials Laboratory I**  
**Fall 2024**

**Instructors:** Jessica TerBush, 207 Ceramics, 217-300-9924, jterbush@illinois.edu

**Lecture:** 1:00 PM – 1:50 PM, Mondays & Wednesdays

***2055 Sidney Lu for FIRST CLASS ONLY***

***122 Kiln House for ALL SUBSEQUENT CLASSES***

**Office Hours:** 1:00 PM – 2:00 PM Tuesdays & Thursdays, or by appointment

**Laboratory:** 2:00 PM – 5:00 PM (M-F) – 105-108 Kiln House

**Course Objectives:**

- 1) To practice the technical skills required to properly use materials science instruments;
- 2) To apply the principles of materials science and engineering through lab investigation;
- 3) To analyze lab results and organize them into logical, concise, and accurate reports;
- 4) To practice and refine writing and communications skills for effective presentation of technical materials.

**Text:** Handouts for individual experiments.

**On reserve in Grainger:** “A Guide to Writing as an Engineer”, Beer and McMurrey, 4th Edition. (Not required, but available for reference.)

**Website for course materials:** <https://canvas.illinois.edu> **and** CampusWire site for Q&A

**Course Outline:**

- |  |      |
|--|------|
| 1. Thermal Stresses and strains (TS)         | 2 Wk |
| 2. Pyrometry (PY)                            | 2 Wk |
| 3. Phase Equilibria: Optical Microscopy (OM) | 2 Wk |
| 4. Absorption and Humidity (AH)              | 2 Wk |
| 5. Thermoelectric Energy Conversion (TEC)    | 2 Wk |
| 6. Differential Scanning Calorimetry (DSC)   | 2 Wk |

## Schedule: Fall 2024

Students will be divided into three groups (A, B, or C), and only need to attend the discussions for their group. Group assignments as of [date] are posted on Canvas.

Week	Group A	Group B	Group C	Week	Monday	Wednesday
1	PY1	OM1	TS1	8/26 – 8/30	Intro	PY-Disc-A
2	PY2*	OM2*	TS2*	9/2-9/6	Labor Day*	OM -Disc-B
3	TS1	PY1	OM1	9/9-9/13	TS -Disc-C	PY-Disc-B
4	TS2	PY2	OM2	9/16-9/20	OM-Disc-C	TS-Disc-A
5	OM1	TS1	PY1	9/23-9/27	PY- Disc-C	OM-Disc-A
6	OM2	TS2	PY2	9/30-10/4	TS- Disc-B	Tech Writ-A
7	<b>Break</b>	<b>Break</b>	<b>Break</b>	10/7-10/11	Tech Writ-B	Tech Writ-C
8	TEC1	AH1	DSC1	10/14-10/18	TEC-Disc-A	DSC-Disc-C
9	TEC2	AH2	DSC2	10/21-10/25	AH-Disc-B	Present-C
10	AH1	DSC1	TEC1	10/28-11/1	TEC-Disc-C	DSC-Disc-B
11	AH2	DSC2	TEC2	11/4-11/8	AH- Disc-A	Present-B
12	DSC1	TEC1	AH1	11/11-11/15	TEC-Disc-B	DSC-Disc-A
13	DSC2	TEC2	AH2	11/18-11/22	AH-Disc-C	Present-A
14	<b>Fall Break</b>	<b>Fall Break</b>	<b>Fall Break</b>	11/25-11/29	<b>No Lecture</b>	<b>No Lecture</b>
15	Presentation	Presentation	Presentation	12/2-12/6	Presentation	Presentation
16	Presentation	Presentation	Presentation	12/9-12/11	Presentation	Presentation

\*Students in the Monday group will need to join one of the other groups on T-F for the week of Labor Day

7HDFKLQJMLMWDQWV

Zirui Wang (ziruiw2@illinois.edu)

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*Undergraduate Teaching Assistants*

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**Lab Manager:** Nicole Robards

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212 Ceramics (4-7498)

## Grading Policies and Procedures:

1. Each student will submit **five (5) individually written reports** during the course of the semester and also will make **an oral presentation** at the end of the semester.
  - **Two** of the reports are formal **Long** reports, **one** of which will have a rewrite opportunity.
  - **Three** are **Short** reports;
2. An **electronic version** of each lab report will be submitted in MSWord or PDF format to the Canvas course website.
3. The reports are due **exactly at 2:00 PM** in the lab, **one week after** the experiment is finished. Any excuses must be presented to the instructor in writing **before** the due date of the report, and only then will the excuse be considered.
4. **All unexcused late reports will be penalized.** Typical penalty is 10% deduction for reports submitted within 24 hours of deadline, but may be higher for reports submitted after this.
5. Each report will be graded using a standard checklist for a total of **100 points**.
  - 30% of which is allocated to the writing (organization, format, grammar, spelling, sentence construction, style and illustration).
  - 70% is allocated to the technical content (concepts, data analysis, interpretation and understanding).
6. Each student is required to maintain a **lab notebook**. The notebook must be **signed and dated** by the TA at the end of each lab to indicate that the student has completed the lab. The student then needs to scan the signed page and submit it to Compass to receive the credit for completing the lab. Completing all 6 labs will add **20%** toward the final grade; however, missing one lab will result in an **incomplete** grade until made up.
7. Everyone is required to make a **15-minute presentation** on the DSC lab. The presentation will be followed by 5-minute questions and answers (Q&A). The presentation will be graded on completion of lab requirements (40%), data analysis (30%), clarity of presentation (15%) and Q&A (15%). The presentations will be held during the last two weeks of the semester. Signup sheets will be posted for scheduling individual presentations.
8. Each lab will have a **pre-lab quiz**, which must be completed online (on the Canvas course website) before the start of the lab. The quizzes will contribute 5% toward the final grade.
9. The final grade will be calculated as following: **60%** from the 5 lab reports (**10%** for each short lab report, **15%** for each long report), **20%** for completing all 6 labs (signatures), **15%** from the presentation, and **5%** from the 6 pre-lab quizzes.
10. Everyone will have the opportunity to **rewrite** the report for the **OM** lab. Rewrites will be due at **11:59 PM on 12/1**. The grade of the initial report will count 40%, while the grade of the rewritten report will add 60% toward the final grade for the OM lab report. A copy of the original report must be submitted with the rewrite.
11. **Final letter grades** will be awarded depending on the **class average** and the **relative performance** of the individuals. Overall, scores **less than 50%** are a **failing grade**.

## Penalties:

- **Copying or Sharing:** 25 - 50% penalty for **copying** or **sharing** any part of the individual report
- **Plagiarism:** As per the University Code of Policies and Regulations, the instructor may impose one or more of the following penalties **depending on the severity** of the infraction:
  - A reduced grade for the lab report
  - A "0" for the lab report
  - A reduced grade for the course
  - A failing grade for the course
  - Recommend to the head of the department that the student be suspended or dismissed from the University

## Laboratory Rules and Regulations:

### A) Safety:

- 1) Always use appropriate **safety equipment** and follow proper **safety procedures**
- 2) Always bring your **own Safety Glasses** and wear them during the lab period (***Students will be provided one pair at the start of the semester. If lost, it is your responsibility to purchase a new pair.***)
- 3) Always **wear proper attire** (shirt, **long pants or ankle-length skirt** and **closed toe shoes**).
- 4) **Long hair?** Properly **contain it** so that it will not be hazardous to you and to your fellow classmates and will not damage the very sensitive equipment in the labs
- 5) Inside the labs, **ABSOLUTELY NO:**
  - **Food or drink**
  - **Roller Skates/Blades**
  - **Game playing** (of any sort, particularly on the computers)
- 6) Anyone caught **violating** the above **laboratory rules:**
  - Will be **expelled** from the lab
  - **No compensating lab time** will be given to complete the experiment(s) and
  - Will **not be allowed** back into the labs without a letter from the Head of the Materials Science and Engineering department.
- 7) **Online Lab Safety Training:** Everyone is required to complete the online lab safety training at the DRS website before **finishing** the first lab. You can complete the training at the following website: <http://www.drs.illinois.edu> – select “Laboratory Safety” from the dropdown menu.  
After you finish the training, submit a copy of your certificate to Canvas. Must be completed within the **first two weeks of class**.

## B) General:

- Arrive at the lab (**105-108 Kiln House**) a few minutes before the start of each lab; TA has the discretion to deduct 2% from the completion grade (20% for all 6 labs) for anyone who is more than 15 minutes late for a lab. Anyone who is more than 30 minutes late for a lab will not receive the credit for the lab and will automatically receive an **incomplete** grade.
- You are responsible for the **clean-up** of the common as well as the individual work areas at the end of your lab period
- After you are done with your experiment, you **should not leave** the lab premises **without checking with your TA**
- Pay particular attention to the proper use of equipment and experimental procedures. Use your down time wisely. **No loud and excessive conversations**
- You **should not leave** the area of your experimental set up in the middle of the experiment **without a proper cause and/or permission** from the **lab supervisor**
- Save your data on your own storage media immediately after the experiment is done.

## University Policies to Note (the “fine print”):

**Emergency Situations:** Emergency response recommendations and campus building floor plans can be found at the following website: <https://police.illinois.edu/em/run-hide-fight/>. I encourage you to review this website within the first 10 days of class.

Campus Instructional Facility:

Ceramics Kiln House: <http://police.illinois.edu/dpsapp/wp-content/uploads/2016/02/u0011.pdf>

**Absence Policy:** Your attendance is required at each lab section, and lecture attendance is strongly encouraged for success in this class. If you need to miss lab for a planned absence (religious observance, conference, travel, interview, etc), please contact Dr. TerBush in advance to reschedule your lab. For last minute absences (illness, family emergency, etc), please contact Dr. TerBush as soon as possible to discuss options for make-up. Likewise, if you need an extension on a report due to illness, travel, etc, please contact Dr. TerBush *in writing* prior to the due date to make arrangements.

**Mental Health:** Significant stress, mood changes, excessive worry, substance/alcohol misuse or interferences in eating or sleep can have an impact on academic performance, social development, and emotional wellbeing. The University of Illinois offers a variety of confidential services including individual and group counseling, crisis intervention, psychiatric services, and specialized screenings which are covered through the Student Health Fee. If you or someone you know experiences any of the above mental health concerns, it is strongly encouraged to contact or visit any of the University’s resources provided below. Getting help is a smart and courageous thing to do for yourself and for those who care about you.

- Counseling Center (217) 333-3704
- McKinley Health Center (217) 333-2700
- National Suicide Prevention Lifeline (800) 273-8255
- Rosecrance Crisis Line (217) 359-4141 (available 24/7, 365 days a year)

If you are in immediate danger, call 911.

**Community of Care:** As members of the Illinois community, we each have a responsibility to express care and concern for one another. If you come across a classmate whose behavior concerns you, whether in regards to their well-being or yours, we encourage you to refer this behavior to the Student Assistance Center (217-333-0050 or <http://odos.illinois.edu/community-of-care/referral/>). Based on your report, the staff in the Student Assistance Center reaches out to students to make sure they have the support they need to be healthy and safe.

Further, as a Community of Care, we want to support you in your overall wellness. We know that students sometimes face challenges that can impact academic performance (examples include mental health concerns, food insecurity, homelessness, personal emergencies). Should you find that you are managing such a challenge and that it is interfering with your coursework, you are encouraged to contact the Student Assistance Center (SAC) in the Office of the Dean of Students for support and referrals to campus and/or community resources.

**Student with Disabilities:** To obtain disability-related academic adjustments and/or auxiliary aids, students with disabilities must contact the course instructor as soon as possible and provide the instructor with a Letter of Academic Accommodations from Disability Resources and Educational Services (DRES). To ensure that disability-related concerns are properly addressed from the beginning, students with disabilities who require assistance to participate in this class should apply for services with DRES and see the instructor as soon as possible. If you need accommodations for any sort of disability, please speak to me after class, or make an appointment to see me or see me during my office hours. DRES provides students with academic accommodations, access, and support services. To contact DRES, you may visit 1207 S. Oak St., Champaign, call 217-333-1970, e-mail [disability@illinois.edu](mailto:disability@illinois.edu) or visit the DRES website at <http://www.disability.illinois.edu>. Here is the direct link to apply for services at DRES, <https://www.disability.illinois.edu/applying-services>.

**Disruptive Behavior:** Behavior that persistently or grossly interferes with classroom activities is considered disruptive behavior and may be subject to disciplinary action. Such behavior inhibits other students' ability to learn and an instructor's ability to teach. A student responsible for disruptive behavior may be required to leave class pending discussion and resolution of the problem and may be reported to the Office for Student Conflict Resolution (<https://conflictresolution.illinois.edu>; [conflictresolution@illinois.edu](mailto:conflictresolution@illinois.edu); 333-3680) for disciplinary action.

**Sexual Misconduct Reporting Obligation:** The University of Illinois is committed to combating sexual misconduct. Faculty and staff members are required to report any instances of sexual misconduct to the University's Title IX and Disability Office. In turn, an individual with the Title IX and Disability Office will provide information about rights and options, including accommodations, support services, the campus disciplinary process, and law enforcement options.

A list of the designated University employees who, as counselors, confidential advisors, and medical professionals, do not have this reporting responsibility and can maintain confidentiality, can be found here: [wecare.illinois.edu/resources/students/#confidential](http://wecare.illinois.edu/resources/students/#confidential).

Other information about resources and reporting is available here: [wecare.illinois.edu](http://wecare.illinois.edu).

**Academic Integrity:** The University of Illinois at Urbana-Champaign Student Code should also be considered as a part of this syllabus. Students should pay particular attention to Article 1, Part 4: Academic Integrity. Read the Code at the following URL: <http://studentcode.illinois.edu/>.

Academic dishonesty may result in a failing grade. Every student is expected to review and abide by the Academic Integrity Policy: <https://studentcode.illinois.edu/article1/part4/1-401/>. Ignorance is not an excuse for any academic dishonesty. It is your responsibility to read this policy to avoid any misunderstanding. Do not hesitate to ask the instructor(s) if you are ever in doubt about what constitutes plagiarism, cheating, or any other breach of academic integrity.

**Religious Observances:** Illinois law requires the University to reasonably accommodate its students' religious beliefs, observances, and practices in regard to admissions, class attendance, and the scheduling of examinations and work requirements. Students should complete the [Request for Accommodation for Religious Observances](#) form should any instructors require an absence letter in order to manage the absence. In order to best facilitate planning and communication between students and faculty, students should make requests for absence letters as early as possible in the semester in which the request applies.

# *Guidelines to Preparing* **Laboratory Reports**

Department of Materials Science and Engineering  
University of Illinois at Urbana-Champaign

The following guidelines are for writing formal laboratory reports for the MatSE-307 and 308 laboratory courses taught in the Department of Materials Science and Engineering at the University of Illinois at Urbana-Champaign.

*All laboratory reports should contain the following in order:*

## **Title Page:**

The following information should be centered on the front page: (with no page number)

**Experiment Number**

**Title of the Experiment**

**Author**

**(Lab Partners)**

**Class \_\_\_\_\_, Section \_\_\_\_\_**

**Date Due \_\_\_\_\_**

**Date Received \_\_\_\_\_** (leave blank)

## **Abstract:**

The abstract is to be placed on numbered Page 1, in block style with no paragraph indentation and in **bold** type. It should be no more than **150 to 200 words** in length.

The abstract should be a concise summary of the experiment, containing general statements of the investigation, the methods used, materials tested, and the main results. It should not include procedural details.

*The following sections of the report should start at the top of Page 2:*

## **Introduction and Background:**

The Introduction and Background section should include: (1) statements that clearly define the purpose of the experiment; (2) its significance; (3) background information necessary to understand the concepts, methods, and procedures presented in the subsequent sections. This section should not contain the details of the laboratory procedures and the data analyses.

The purpose of this section is to identify the reasons for performing the experiment. What are you measuring or determining and why? What is the significance of your observations? Say this in your own words. Do not copy the "objectives" from the laboratory manual or any handout that might be given to you. The goal is to supply the minimum supplementary information necessary to understand the methodology employed and the theoretical background of the experiment. When this section is properly written, the logic behind the experimental approach will be clear.

## **Experimental Procedure:**

The Experimental Procedure section should give an explicit and concise account of the methods and procedures followed during the experiment without getting into the operational details of the equipment used. Do not simply copy the laboratory manual or a handout. A person with technical background should be able to carry out the experiment without any difficulty by reading this section and the appropriate equipment manuals.

This section should include, when applicable, raw materials, compositions, batch formulas, and specific processing and analytical procedures used to make, test, and analyze the samples. Brief descriptions of equipment, diagrams of apparatus, sketches of circuits, etc., should be included when needed for clarity.

### **Results and Discussion:**

This section should be the majority of your report. All pertinent observations and the refined data should be presented and discussed in logical order. Whenever possible, data should be presented in graphical form to show any relationships between variables. Raw data should be tabulated and placed in an appendix.

The interpretations and significance of the results should be discussed, including whether the purpose of the experiment had been fulfilled. The results should be compared with those found in the literature and if possible, with theory. Any specific experimental conditions that may have affected the results and any sources of error should also be discussed.

### **Conclusions:**

This section should be a brief summary of the important findings, preferably in itemized form. The conclusions should not simply be a copy and paste of the abstract. Conclusions are a distinct element within a technical report.

### **References:**

References acknowledge the sources of non-original information, data, and ideas (i.e., not those of the author), and are used to support the author's point of view or observations.

References should be numbered consecutively in the order they appear in the text and listed in this section with the following information. The references should be listed in IEEE format as indicated by the following examples:

#### ***Journal:***

<sup>1</sup>R.W. Vest, "Materials Science of Thick Film Technology," *Am. Ceram. Soc. Bull.*, vol. 65 [4], pp. 631-36, 1986.

#### ***Proceedings:***

<sup>2</sup>R. A. Perecherla and R.C. Buchanan, "Copper Thick Film Adhesion on Glass Ceramic Substrates," in *Proceedings of the American Ceramic Society Symposium*, Indianapolis, IN, 1989, pp. 439-454.

#### ***Books:***

<sup>3</sup>W. D. Kingery, H. K. Bowen, and D.R. Uhlmann, "Dielectric Properties" in *Introduction to Ceramics*, 2<sup>nd</sup> Ed., New York, NY, John Wiley and Sons, 1976, pp. 913-74.

#### ***Patents:***

<sup>4</sup>M. J. Pryor and T. J. Gray, "Method of Preparing Molten Metal Filter," US Patent #4,056,586, Nov 1, 1977.

### **Appendices:**

The Appendices are used for placing material that is pertinent, but would cause a disruption to the flow of the text in the main body of the report. Raw data, sample calculations, and extensive derivations are typically included. Appendices should be lettered consecutively (e.g., Appendix A, Appendix B, etc.) and given a descriptive title.



## General Guidelines:

### Format:

All reports must be *typewritten, single sided and double-spaced, with one-inch margins on all sides and the pages numbered*. Reports must be either stapled at the upper left-hand corner or bound in a report cover. Any handwritten annotations, equations, etc. must be neat and legible.

A good laboratory report should be concise and comprehensive, and need not be long. The *text* of a typical laboratory report usually ranges from *10 to 13 pages* in length (excluding tables and figures).

### Writing:

- *Always save your report file on a backup disk*
- Use concise and economical wording of sentences and paragraphs without being choppy
- Particular attention should be given to proper word usage, correct spelling, proper punctuation, and the use of complete sentences
- Keep the tense consistent in a given section. Usually simple present tense is best. The main exception is the description of the experimental procedure where past tense is often more natural.
- Typically, sentences should be written in **third person**, but **first person** can be used where appropriate. Avoid a conversational style of writing.
- Avoid using gender-specific references.
- Avoid the use of absolute, unrestricted, or unqualified statements
- Reports should be written based on the premise that the reader may not be well informed on the subject. Reports should be concise, but they must be sufficiently complete to supply the necessary information.
- If you have general comments, which you care to make about the laboratory, this is not the place to make them. Such comments should be directed to the instructor or to a teaching assistant on a separate note.

### Tables:

All tables should be numbered in the order they appear and placed successively in the text. Each table should have a descriptive caption *above* the table, and numbered as Table I, Table II, etc. as per the following example:

Table I. Mg and Fe Contents in the External Oxide Layer as a Function of Time and Temperature

Sample no.	Oxidation temperature (°C)	Oxidation time (h)	Fe content ( $\times 10^{15}/\text{cm}^2$ )	Mg Content ( $\times 10^{15}/\text{cm}^2$ )
4-1a	1000	0.5	120	160
4-1b	1000	1.0	210	190
4-1c	1000	4.0	320	340
4-1d	1000	10.0	480	550
4-1g	1000	40.0	960	1200
5-b1	700	10.0	33	29
5-b3	800	10.0	100	100
2-1a	900	10.0	330	460
5-b2	1000	10.0	470	480
5-b5	1100	10.0	1100	770
10-b3b	700	70.0	220	150
10-b4b	810	49.3	460	250
10-b5b	907	10.0	440	310
9-b5b	1100	1.0	350	160

**Figures:**

All figures should also be numbered in the order they appear and placed successively in the text. Each figure should have a descriptive caption *below* the figure, and numbered as Fig. 1, Fig. 2, etc. as per the following examples:

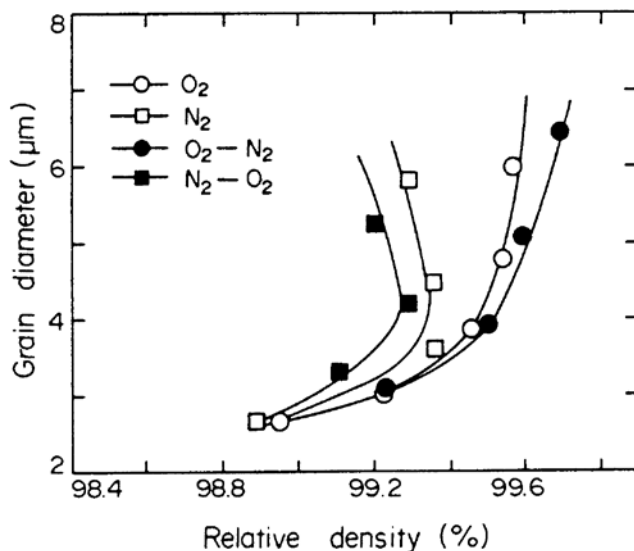


Fig. 1. Acceptable example of a graph.

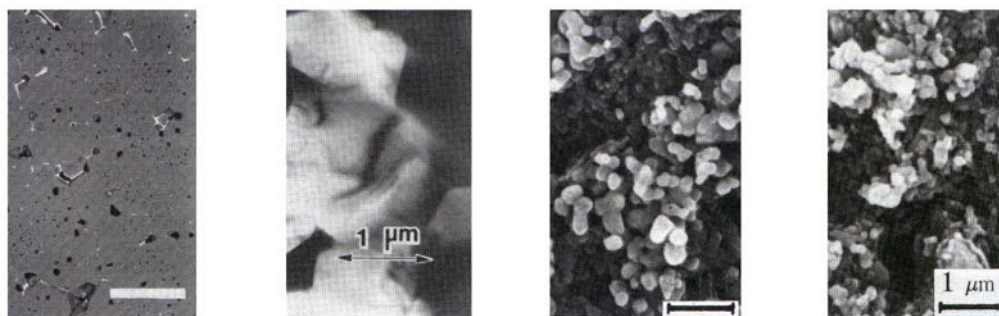


Fig. 2. Acceptable examples of scale bars.

When micrographs are to be included in the report, use scale bars as shown above and indicate the magnification in the caption, in parenthesis.

The axes of each graph must be labeled and their units clearly designated in parenthesis. Take error into account.

**Mathematical Expressions:**

Equations should be consecutively numbered and all variables identified. Decimal numbers less than one must have a leading zero such as 0.01, not .01. Every symbol must be defined, and avoid multiple meanings for the same symbol. In displaying equations with fractions (except superscripts), numerators should be stacked over denominators rather than placed on a single line and separated by a slash (/). Very large and very small numbers should be expressed in scientific notation, e.g.,  $4.53 \times 10^8$  and  $2.98 \times 10^{-8}$ .

**Note:**

If you choose to include tables and figures within the body of your report, always place them (along with the equations) after their point of reference. Otherwise, put your tables and figures at the end of the report (after References)

**SI Units:**

Use metric units of measurement, specifically SI units. English or non-metric units may appear in the report, but they should appear only in parenthesis following the SI units, e.g., 32 mm (1.25 in.).

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## **Formal (Long) Reports**

*The long reports should contain the following sections in order:*

- 1. Title Page**
- 2. Abstract**
- 3. Introduction and Background**
- 4. Experimental Procedure**
- 5. Results and Discussion**
- 6. Conclusions**
- 7. References**
- 8. Appendices**

## **Short Reports**

*The short reports should contain the following sections in order:*

- 1. Title Page**
- 2. Abstract**
- 3. Objectives and Experimental Procedure (What? and How?)**  
Maximum of **one page** in length  
(This section should contain the main objectives of the experiment and a brief experimental procedure.)
- 4. Results and Discussion**
- 5. Conclusions**
- 6. References**  
(**DO NOT** include any Appendices)

## Type of Reports / MSE-307

<u>Experiment</u>	<u># of weeks</u>	<u>Report Type</u>
1. Pyrometry / (PY)	2 Wks	Short
2. Thermal Stresses and Strains / (TS)	2 Wks	Short
3. Phase Equilibria: Optical Microscopy / (OM)	2 Wks	Long (rewrite)
4. Absorption and Humidity / (AH)	2 Wks	Short
5. Thermoelectric Energy Conversion / (TEC)	2 Wks	Long
6. Differential Scanning Calorimetry / (DSC)	2 Wks	Presentation

### **Presentation Guidelines**

Location: TBA, likely 122 KH

Format: Electronic presentation in Powerpoint format

Duration: 15 minutes + Q&A

Date: M-F afternoons during the last 1.5 weeks of class

Everyone should sign up for one 20-minute presentation slot. Sign-up will be done electronically through Google Docs (watch email for link).

Please upload a copy of your slides (.ppt or .pdf) to Compass.

A group of two instructors (TAs, UG assistants, and/or Dr. TerBush) will grade the presentations. The presentation grade will be determined by the following four components:

- 1) Completion of lab requirements (40%)
- 2) Data analysis (30%)
- 3) Clarity of presentation (15%)
- 4) Q&A (15%).